



UNITED STATES ENVIRONMENTAL PROTECTION AGENCY
CENTER FOR ENVIRONMENTAL MEASUREMENT AND MODELING
RESEARCH TRIANGLE PARK, NC 27711

OFFICE OF
RESEARCH AND DEVELOPMENT

December 2, 2019

Mr. Steven E. Flint, Director – Division of Air Resources
New York State Department of Environmental Conservation (NYSDEC)
625 Broadway
Albany, New York 12233-3250

Dear Mr. Flint:

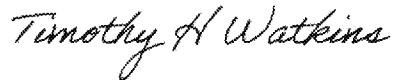
In response to your July 20, 2018 request for assistance to qualitatively identify per- and polyfluoroalkyl substances (PFAS) in the emissions from an industrial source that is thermally processing polytetrafluoroethylene (PTFE) powders, I am pleased to provide the enclosed report. The report provides qualitative targeted and non-targeted analytical results identifying and tentatively identifying various PFAS and PTFE thermal degradation products from process emission samples. These samples were collected by the U.S. Environmental Protection Agency's (EPA's or the Agency's) Office of Research and Development (ORD) in ongoing collaborative technical support to the New York State Department of Environmental Conservation (NYSDEC) at a PTFE powder thermal processing facility.

NYSDEC's request was to aid in the investigation to the presence of PFAS in the environment near manufacturing facilities of interest. This related well to ORD's research capabilities and interests to apply qualitative targeted and non-targeted sampling and analysis methods for discovery of the nature and extent of PFAS environmental occurrence associated with industrial releases. ORD continues to develop emissions sampling and analytical methods for many PFAS compounds in various media, including some of those included in the report.

The report provided does not interpret exposure or risk from the tentative identification of PFAS compounds and PTFE thermal degradation products. EPA does not currently have health-based standards, toxicity factors, or associated risk levels for PFAS, other than perfluorooctanoic acid (PFOA), perfluorooctane sulfonate (PFOS), and perfluorobutanesulfonic acid (PFBS). While the report data indicate the presence (or lack) of specific PFAS in the process emissions, we do not have sufficient information to offer interpretations related to human or environmental exposure, or risk.

Thank you for inviting us to be part of this effort to further both EPA's and NYDEC's understanding of this important issue. This is just one of many Agency efforts that demonstrates EPA's commitment to cooperative federalism. If you have any questions or concerns, do not hesitate to contact me at (919) 541-2106 or via email at [[HYPERLINK "mailto:watkins.tim@epa.gov" \h](mailto:watkins.tim@epa.gov)] or Lara Phelps at (919) 541-5544 or via email at [[HYPERLINK "mailto:phelps.lara@epa.gov" \h](mailto:phelps.lara@epa.gov)] I look forward to our continued work together.

Sincerely,



Timothy H. Watkins
Director

Enclosure

CC: Peter Lopez, USEPA, Region 2
Walter Mugdan, USEPA, Region 2
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Doug Garbarini, USEPA, Region 2
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**Joint ORD/New York State Department of Environmental Conservation:
Saint-Gobain PTFE Sintering Facility Emissions Characterization Study**

Study Description

The New York State Department of Environmental Conservation (NYSDEC), Division of Air Resources (DAR), and EPA Region 2 have requested the assistance of EPA's Office of Research and Development (ORD) in performing targeted and non-targeted Poly- and Perfluoroalkyl Substances (PFAS) emissions analyses to qualitatively identify air emissions from process source operations. Specifically, ORD has been requested to qualitatively identify various PFAS and other thermal degradation products in stack emissions from the polytetrafluoroethylene (PTFE) process. Quantitative analyses are not within the scope of this study.

The process studied is an industrial source where bulk PTFE (~1,000 lb) is molded and then sintered in an oven. The sintering process involves a gradual heating of the PTFE billet until it reaches the desired sintering temperature (~700 °F). The overall sintering process requires ~5 days, including ~2 days for cooling. Potential emissions include residual PFAS as well as thermal degradation products, such as PTFE monomers.

The technical objectives of this joint ORD – NYSDEC study were to characterize the emissions, as comprehensively as possible, from the PTFE sintering process. This includes characterizing emissions as a function of sintering process time and temperature using targeted and non-targeted measurement approaches. Investigating the potential presence of PFOA is the primary targeted measurement of interest, as well as the presence of thermal degradation products such as tetrafluoroethylene (TFE), hexafluoropropylene (HFP), and perfluoroisobutylene (PFIB). Non-targeted measurements were conducted to tentatively identify potential additional process emission compounds not identified by targeted measurement approaches.

This data report is intended to provide a simple representation and summary of the study results. Therefore, the description of methods and quality assurance are brief and high-level. In this report, we provide identification and tentative identification of PFAS compounds and PTFE thermal degradation products. We do not interpret exposure or risk from these values. EPA does not currently have health-based standards, toxicity factors, or associated risk levels for PFAS, other than perfluorooctanoic acid (PFOA), perfluorooctane sulfonate (PFOS), and perfluorobutanesulfonic acid (PFBS). While the data provided in this report indicate the presence (or lack) of specific PFAS in the process emissions, we do not have sufficient information to offer interpretations related to human or environmental exposure and risk.

ORD personnel were responsible for all on-site sample collection and measurements, as well as all analyses performed. ORD's sampling, analysis, and report team that contributed to this effort are listed in Table 1.

Table 1. EPA Office of Research and Development sampling, analysis, and report team.

Responsibility	Personnel
Project Technical Lead	Jeff Ryan
Emissions sampling	Jeff Ryan, Ken Krebs, Theran Riedel, John Offenberg
Laboratory chemistry	James McCord, Ingrid George
Quality Assurance Review	Libby Nessley, Bob Wright
Management coordination and review	Lara Phelps
Report Preparation	Jeff Ryan, Ken Krebs

Methods in Brief

Sampling Approach:

Multiple gaseous emissions samples were collected at the sintering process uncontrolled exhaust. Emissions samples were collected as a function of sintering process time and temperature over the course of ~70h of emissions testing.

Whole air samples were collected with SUMMA[®] canisters for USEPA Method TO-15 (EPA/625/R-96/010b, January 1999) targeted qualitative analyses using gas chromatography/mass spectrometry (GC/MS), targeted qualitative thermal degradation products analyses (GC/MS), and qualitative non-targeted PFAS screening analyses, also using GC/MS.

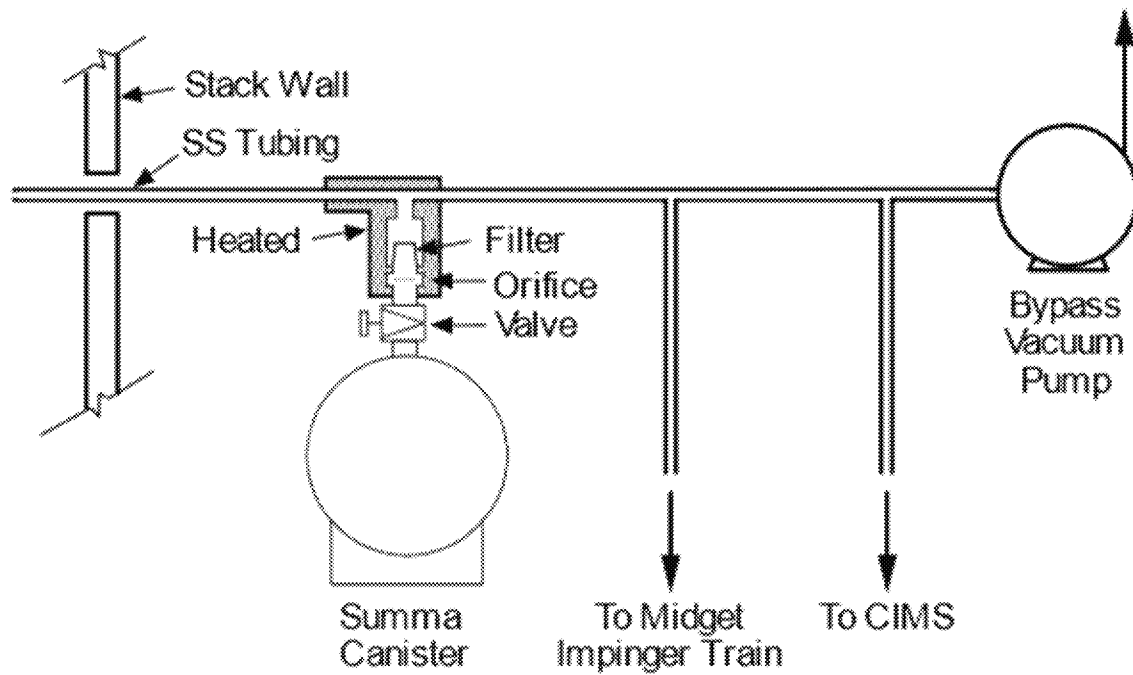
An aqueous, four-impinger, midjet impinger train was used to capture hydrophilic PFAS for targeted and non-targeted, qualitative PFAS analyses. This approach focuses on the qualitative capture of hydrophilic PFAS compounds, including PFOA, that may be present in the PTFE sintering process exhaust and is not intended to quantitatively capture all potential PFAS compounds.

ORD's iodide adduct high resolution chemical ionization mass spectrometry (CIMS) was used as a real-time, on-line process characterization monitor. The CIMS is capable of detecting specific polar compounds such as polyfluorinated carboxylic acids (PFCAs), including PFOA, as well as fluorotelomer alcohols (FTOHs).

All samples were collected from a sampling manifold where the process emissions are extracted from the stack with a dedicated sampling system. The sampling system, depicted in Figure 1, consists of an unheated sample probe and heated tubing from the probe to the heated manifold. The emissions sample was withdrawn from the stack with a bypass vacuum pump. All emissions samples were collected from the heated manifold positioned on the vacuum side of the sample pump.

A total of 16 - 2h SUMMA[®] and impinger train emissions samples were collected over the course of the ~70h oven heating cycle. No emissions samples were collected during the oven cooling cycle. A single ambient SUMMA sample was collected indoors in an area in the vicinity of the sintering ovens. Prior to initiation of the oven cycle, a system blank was performed where single SUMMA and impinger samples were collected while nitrogen was introduced at the sample probe and routed through the entire sampling system in the same manner as process emissions as a means to assess potential overall system contamination.

Figure 1. Sampling System



Analytical Approach:

Targeted GC/MS analysis of the SUMMA[®] canisters for TO-15 VOCs was conducted on a GC coupled to a quadrupole mass spectrometer. Samples were analyzed under electron ionization selective ion monitoring (SIM), MS conditions. Chromatographic 'area under the curve' are reported for each identified compound. No internal standard masses (i.e. 'mass added') nor peak areas are reported, even if added to these samples. As a result, these targeted analyses are qualitatively only. The TO-15 target analyte list is presented in Table 2.

Table 2. TO-15 Target VOCs

Propylene	Methylene Chloride	2-Methylhexane	Bromoform
Propane	3-Chloro-1-Propene	2,3-Dimethylpentane	Styrene
Dichlorodifluoromethane	1,1,2-Trichloro-1,2,2-...	Tert Amyl Methyl Ether	1,1,2,2-Tetrachloroethane
Chloromethane	Carbon Disulfide	3-Methylhexane	o-Xylene
Isobutane	2,2-Dimethylbutane	1,2-Dichloropropane	Nonane
Dichlorotetrafluoroethane	trans-1,2-Dichloroethene	Bromodichloromethane	Bromofluorobenzene
Vinyl Chloride	Cyclopentane	1,4-Dioxane	Chlorotoluenes
1-Butene	2,3-Dimethylbutane	Trichloroethene	n-Propylbenzene
1,3-Butadiene	1,1-Dichloroethane	Isooctane	m-Ethyltoluene
Butane	Methyl-t-Butyl-Ether	Methyl Methacrylate	1,3,5-Trimethylbenzene
trans-2-butene	Vinyl Acetate	Heptane	1,2,4-Trimethylbenzene
Bromomethane	2-Methylpentane	cis-1,3-Dichloropropene	Tert-Butyl Benzene
cis-2-butene	2-Butanone	4-Methy-2-Pentanone	1-Ethyl-4-Methyl Benzene
Chloroethane	3-Methylpentane	Methylcyclohexane	o-Ethyltoluene
Ethanol	2-Chloroprene	trans-1,3-Dichloropropene	1,3-Dichlorobenzene
Vinyl Bromide	1-Hexene	1,1,2-Trichloroethane	1,4-Dichlorobenzene
Acetonitrile	cis-1,2-Dichloroethene	2,3,4-Trimethylpentane	n-Decane
Acrolein	Diisopropyl ether	Toluene	Sec-Butyl Benzene
Acetone	Ethyl Acetate	2-Methylheptane	1,2,3-Trimethylbenzene
iso-Pentane	n-Hexane	2-Hexanone	1,2-Dichlorobenzene
Trichlorofluoromethane	Chloroform	Dibromochloromethane	o-Cymene
Isopropyl Alcohol	Tetrahydrofuran	3-Methylheptane	1,3-Diethylbenzene
1-Pentene	Ethyl Tert-Butyl Ether	1,2-Dibromoethane	1,2-Diethylbenzene
Acrylonitrile	Methylcyclopentane	Octane	n-Butyl Benzene
n-Pentane	1,2-Dichloroethane	Tetrachloroethene	Undecane
Isoprene	2,4-Dimethylpentane	1,1,1,2-Tetrachloroethane	1,2,4-Trichlorobenzene
trans-2-pentene	1,1,1-Trichloroethane	Chlorobenzene	Naphthalene
cis-2-pentene	Benzene	Ethylbenzene	Dodecane
Tert-Butanol	Carbon Tetrachloride	m-Xylene	Hexachlorobutadiene
1,1-Dichloroethene	Cyclohexane	p-Xylene	

Targeted GC/MS analysis of the SUMMA[®] canisters for TFE, HFP, and PFIB, as well as 4:2, 6:2, and 8:2 telomer alcohols, were conducted on a GC coupled to a quadrupole, time-of-flight mass spectrometer (TOF/MS). Samples were analyzed under electron ionization full-scan, MS conditions. Gas standards for TFE, HFP, and the telomer alcohols were used to aid in compound identification/compound absence. Chromatographic 'area under the curve' are reported for each identified compound. No internal standard masses (i.e. 'mass added'), nor peak areas are reported even if added to these samples. As a result, these targeted analyses are qualitatively only. As no gas standard for PFIB can be used, tentative compound identification/compound absence was determined with the aid of reference spectra and retention indices.

Originally, the intent was to perform both targeted and non-targeted liquid chromatography-mass spectrometry (LC-MS) analyses on the impinger samples. Targeted analyses were not performed and deemed not necessary due to several factors. The LC/MS system used for targeted PFAS analyses was not available during the time period of this study. Since this study is limited to qualitative analyses, targeted PFAS analyses for purposes of quantification were not necessary. The LC/MS used for non-targeted analyses was also more powerful and will detect/identify the targeted PFAS as part of the non-targeted analyses.

The non-targeted PFAS analyses were performed on an ultra-performance liquid chromatograph (UPLC) system, coupled to a high resolution, high mass accuracy mass spectrometer. Where possible, PFAS compounds are positively identified against PFAS analytical standards. However, the availability of PFAS analytical standards is limited and often inadequate to identify the PFAS present in many environmental samples. The primary method for tentative PFAS identification is based on the interpretation of mass spectral data – molecular mass along with patterns of fragmentation.

Spectral information is compared to mass-spectral libraries (such as the *U.S. EPA CompTox Chemicals Dashboard* <https://comptox.epa.gov/dashboard>) to tentatively identify compounds present. In addition, PFAS compounds are tentatively identified using specialized spectral interpretation software and analyst expertise. PFAS compounds are tentatively identified to various levels of confidence, depending on the level of combined evidence from manual examination of MS/MS fragmentation spectra and/or comparison with mass spectral libraries. This approach is further described in McCord et al. 2019.¹

The mass spectrometer detector provides integrated peak areas for each compound within the chromatogram. The peak area counts are proportional to the mass of PFAS in the sample. However, without a standard we are not able to derive a mass or concentration value, and results are considered qualitative. As a result, only peak area counts (responses) are reported and these non-targeted analyses are qualitative only.

¹ McCord, J., Strynar, M. Identifying Per- and Polyfluorinated Chemical Species with a Combined Targeted and Non-Targeted-Screening High-Resolution Mass Spectrometry Workflow. *J. Vis. Exp.* (146), e59142, doi:10.3791/59142 (2019). <http://www.jove.com/video/59142/identifying-per-polyfluorinated-chemical-species-with-combined> \h [ve] <http://www.jove.com/video/59142/identifying-per-polyfluorinated-chemical-species-with-combined> \h [m] <http://www.jove.com/video/59142/identifying-per-polyfluorinated-chemical-species-with-combined> \h [j] <http://www.jove.com/video/59142/identifying-per-polyfluorinated-chemical-species-with-combined> \h]

Quality Assurance/Quality Control Summary

This study adhered to the associated Quality Assurance Project Plan (QAPP) (*Joint ORD/New York State Department of Environmental Conservation: Saint-Gobain PTFE Sintering Facility Emissions Characterization Study QA Track ID G-AEMD-0031868-QP-1-0*) prepared and approved prior to testing. The only deviation from the original study plan was that targeted PFAS analyses were not performed on the impinger samples. Relying solely on the non-targeted PFAS analyses does not limit reported data quality in any way. The justification for this has been described above.

The collection and analysis of a variety of reagent and method blanks to account for any PFAS contamination that may have occurred during sampling and analysis was integral to the interpretation and validation of reported data. Multiple blank samples were collected and analyzed, including trip, field, laboratory, instrument, and system blanks. The primary purpose of these blanks is to isolate what is attributable to the samples due to contamination, and to ensure that the data reported are based on actual sintering process emissions. All blank measurements are reported along with emissions data.

The compounds identified and reported are based on their relative relationship to the various blanks and their measured levels. As the primary means of data validation, only those compounds with area counts/responses ≥ 10 times the associated blank level are considered as present in process emissions. There are a total of 16 emissions samples. A compound is considered present if it is found at ≥ 10 times the associated blank level for even a single sample. However, area counts for all 16 samples are reported. The peak area data for determining these relative ratios are contained in this report.

Summary of Results

The qualitative SUMMA[®] results for the targeted TO-15 analyses are reported in Table 3. The compounds identified and the associated area counts are reported. A total of 39 of the 119 VOC compounds targeted for analysis were present in process emissions. The ≥ 10 times the associated blank level is relative to the nitrogen system blank sample.

The qualitative SUMMA[®] results for the targeted TFE, HFP, PFIB, and the 4:2, 6:2, 8:2 telomer alcohol analyses are reported in Table 4. The compounds identified and the associated area counts are reported. Only TFE and 4:2 telomer alcohol were present in process emissions. HFP and PFIB were not present in process emissions. The ≥ 10 times the associated blank level is relative to the nitrogen system blank sample.

The CIMS was operated continuously from the start of process oven heating cycle until the end of the heating cycle, when the cooling cycle began. One-minute data collection averages were electronically stored. The CIMS responses over the entire heating cycle are graphically depicted in Figure 2. Only the C2 (perfluoroethanoic acid), C3 (perfluoropropanoic acid), and C4 (perfluorobutanoic acid) perfluorocarboxylic acids (PFCAs) were present in process emissions. It should be noted that the relative responses to each other should not be interpreted as concentrations relative to each other, since the absolute CIMS response varies among individual compounds.

The impinger results for the non-targeted PFAS analyses are presented in Tables 5-7. The compounds tentatively identified and the associated area counts are reported. A total of 15

polyfluoro PFAS compounds were tentatively identified. The presence of these PFAS compounds in process emissions was limited to sample 1 only. This indicates that these compounds were emitted very early in the oven heating process. The ≥ 10 times the associated blank level is relative to the nitrogen system blank sample. With the exception of the tentative identification of (3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-Heptadecafluorodecyl) hydrogen maleate and 6:2 fluorotelomer phosphate diester (which were diagnostically matched against the U.S. EPA CompTox Chemicals Dashboard), only the molecular mass and formulae can be established. Actual PFAS compound identities cannot be established.

PFOA was conclusively not present in any emissions. No other “legacy” PFAS compounds commonly targeted for analysis (e.g., EPA Method 537 compounds such as PFOS, PFBS, etc) were found in emissions samples.

Table 5 presents a summary of the results from the multiple blanks that were collected and integral to interpreting the impinger data. Table 6 presents the results from emission samples 1, 3, 5, 7, 9, 11, 13, and 15. Table 7 presents the results from emission samples 2, 4, 6, 8, 10, 12, 14, and 16. These data are separated to aid in the interpretation of results. These results indicate that the PFAS compounds tentatively identified were only present in sample 1. Although PFAS were also detected in samples 3, 5, 7, 9, 11, 13, and 15, this is a result of equipment train contamination (see below). PFAS compounds are conclusively not present in samples 2, 4, 6, 8, 10, 12, 14 and 16, and their absence in sample 2 indicates that the 15 PFAS tentatively identified are only present in the process emissions collected by sample 1. As a result, these 15 PFAS compounds are emitted very early in the oven heating cycle.

The fact that the 15 PFAS compounds were present in samples 3, 5, 7, 9, 11, 13, and 15 is due to contamination of the Train 2 glassware (impingers) following the collection of sample 1. The impinger samples were collected with two separate and clearly identified sets of glassware. Samples 1, 3, 5, 7, 9, 11, 13, and 15 were collected with Train 2 and samples 2, 4, 6, 8, 10, 12, 14, and 16 were collected with Train 1. Both sets of glassware were confirmed to be relatively free of contaminants prior to sampling (the proof blanks). The subsequent field blanks (blanks from glassware used for sampling) indicate a contaminant carryover in Train 2 only.

In summary, this study has characterized the emissions from the subject PTFE sintering process. Sixteen separate, 2h emissions samples were collected over the course of the ~70h heating process for TO-15 VOCs, PTFE thermal degradation products, and hydrophilic PFAS compounds. The CIMS, a real-time measurement, also characterized select PFAS compounds as a function of time over the complete heating process. The qualitative, targeted measurements from the SUMMA samples identified 39 TO-15 VOC compounds and 2 thermal degradation products (TFE and 4:2 telomer alcohol) present in process emissions. HFP and PFIB were not present in process emissions. The CIMS results identified the presence of C2, C3 and C4 PFCAs. Non-targeted analyses of the impinger samples detected and tentatively identified 15 polyfluorinated compounds present in process emissions. It was determined that these PFAS compounds were only present in sample 1, indicating that they were emitted very early in the heating process. PFOA was not identified in any of the 16 impinger process emissions samples.

Table 3. Qualitative TO-15 Results from SUMMA® Canisters – Compounds Identified with Associated Area Counts

Target Compounds	Inst Blank	Inst Blank	Trip Blank	Field Blank	Ambient	System Blank	Sample 1	Sample 2	Sample 3	Sample 4	Sample 5
	ppb	ppb	ppb	ppb	ppb	ppb	ppb	ppb	ppb	ppb	Ppb
TFE	ND	ND	0.04	0.01	0.03	0.05	0.13	0.05	0.03	0.05	0.03
Propylene	0.00	0.00	0.00	0.69	19.59	0.47	17.06	196.93	56.18	50.19	42.46
Propane	0.00	0.00	0.00	1.81	74.93	1.44	59.41	699.78	206.52	186.79	153.76
Dichlorodifluoromethane	0.00	0.00	0.00	0.00	0.40	0.00	0.40	0.48	0.54	0.46	0.45
Butane	0.00	0.00	0.00	0.04	0.30	0.09	0.74	1.39	0.66	3.01	3.19
Ethanol	0.00	0.00	0.00	137	666	76.4	1490	2660	2120	2710	2800
Acetone	0.00	0.00	0.00	0.00	5.15	0.00	7.08	7.22	7.30	14.1	27.5
iso-Pentane	0.00	0.00	0.00	0.03	0.77	0.00	0.84	0.68	0.99	1.70	2.24
Trichlorofluoromethane	0.00	0.00	0.00	0.00	0.24	0.00	0.25	0.26	0.28	0.30	0.34
Isopropyl Alcohol	0.00	0.00	0.00	0.44	1.94	0.46	3.38	12.84	13.89	4.37	9.25
1-Pentene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.20	1.27	0.24	0.28
n-Pentane	0.00	0.00	0.00	0.06	0.84	0.08	0.98	0.60	1.23	1.78	1.56
Isoprene	0.00	0.00	0.00	0.00	0.15	0.00	0.06	0.08	0.15	0.17	0.15
trans-2-pentene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Methylene Chloride	0.00	0.00	0.00	0.00	0.08	0.00	0.07	0.21	0.18	0.26	0.33
2-Chloroprene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.23	0.00
1-Hexene	0.00	0.00	0.00	0.00	0.05	0.00	0.00	0.07	0.00	1.65	0.38
Diisopropyl ether	0.00	0.00	0.00	0.05	0.26	0.00	2.44	4.21	2.44	5.79	6.69
Ethyl Acetate	0.00	0.00	0.00	0.00	0.00	0.00	13.0	24.0	13.0	32.7	37.0
n-Hexane	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.45	0.64	1.38	1.61
Methylcyclopentane	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.11	0.00	0.00	0.28
2,4-Dimethylpentane	0.00	0.00	0.00	0.00	0.00	0.00	0.18	0.32	0.18	0.67	0.87
Benzene	0.00	0.00	0.00	0.00	0.16	0.00	0.12	0.19	0.14	0.19	0.21
Carbon Tetrachloride	0.00	0.00	0.00	0.00	0.05	0.00	0.07	0.07	0.09	0.10	0.10
2-Methylhexane	0.00	0.00	0.00	0.12	0.12	0.10	2.39	4.55	2.47	9.73	12.42
3-methylhexane	0.00	0.00	0.00	0.11	0.10	0.13	2.39	4.83	2.57	10.4	12.9
Trichloroethene	0.00	0.00	0.00	0.00	0.11	0.00	0.07	0.07	0.13	0.17	0.19
Heptane	0.00	0.00	0.00	0.00	0.00	0.00	0.25	0.55	0.34	1.22	1.42
Methylcyclohexane	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.23	0.00
Toluene	0.00	0.00	0.00	0.00	62.7	0.00	42.2	6.87	2.07	1.56	1.08
Ethylbenzene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
m-Xylene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.13	0.00	0.00
o-Xylene	0.00	0.00	0.00	0.00	0.00	0.00	0.07	0.09	0.26	0.27	0.30
Chlorotoluenes	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
n-Propylbenzene	0.00	0.00	0.00	0.00	0.56	0.00	0.00	0.00	0.00	0.00	0.00
n-Decane	0.00	0.00	0.00	0.00	0.00	0.00	0.88	20.79	37.71	44.70	75.63
1,2,3-Trimethylbenzene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.04	0.06	0.07	0.00
Undecane	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.13	0.11	0.00
Naphthalene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.08	0.29	0.11	0.00
Dodecane	0.00	0.00	0.00	0.69	19.6	0.00	0.11	0.32	0.49	0.15	0.21

Table 3. Qualitative TO-15 Results from SUMMA® Canisters – Compounds Identified with Associated Area Counts (cont.)

	Sample 6	Sample 7	Sample 8	Sample 9	Sample 10	Sample 11	Sample 12	Sample 13	Sample 14	Sample 15	Sample 16
Target Compounds	ppb	ppb	ppb	ppb	ppb	ppb	ppb	ppb	ppb	ppb	Ppb
TFE	0.05	0.05	0.08	0.05	0.10	0.13	0.57	0.52	0.51	0.32	0.54
Propylene	47.2	35.0	50.5	48.0	47.0	32.6	21.8	24.9	39.5	25.7	40.4
Propane	170	115	163	163	163	121	74.8	83.8	133	86.8	140
Dichlorodifluoromethane	0.45	0.40	0.47	0.50	0.48	0.00	0.48	0.41	0.44	0.44	0.44
Butane	2.49	2.10	1.47	1.43	0.78	0.00	0.74	1.63	2.31	0.86	0.73
Ethanol	2440	3040	2790	2510	2480	2110	2390	1870	2360	1200	1367
Acetone	21.7	32.0	56	30.1	22.1	15.2	6.92	11.3	15.5	6.12	7.96
iso-Pentane	2.49	3.04	4.12	2.85	2.40	1.52	0.96	1.81	2.91	1.14	1.53
Trichlorofluoromethane	0.37	0.33	0.30	0.27	0.27	0.24	0.25	0.28	0.25	0.25	0.23
Isopropyl Alcohol	8.29	19.4	14.4	31.4	14.3	22.0	2.57	9.07	19.8	6.05	19.3
1-Pentene	0.23	0.42	0.58	0.43	0.42	0.28	0.24	0.21	0.33	0.60	0.00
n-Pentane	2.10	2.19	2.40	2.06	2.45	1.21	0.75	1.84	2.69	1.54	1.86
Isoprene	0.18	0.17	0.23	0.33	0.27	0.17	0.08	0.07	0.35	0.07	0.26
trans-2-pentene	0.00	0.06	0.10	0.13	0.06	0.00	0.00	0.00	0.09	0.00	0.00
Methylene Chloride	0.21	0.33	0.23	0.13	0.19	0.13	1.26	0.16	0.15	0.12	0.14
2-Chloroprene	0.00	0.00	0.00	0.07	0.00	0.00	0.00	0.00	0.00	0.00	0.00
1-Hexene	0.12	0.56	0.22	0.19	0.13	0.08	0.06	0.07	0.11	0.09	0.12
Diisopropyl ether	4.04	8.06	4.61	2.34	2.22	1.22	1.40	1.63	2.08	0.50	0.53
Ethyl Acetate	21.6	44.2	25.9	12.4	11.6	6.37	7.47	8.26	11.3	3.78	4.10
n-Hexane	1.49	1.11	0.63	0.38	0.93	0.00	0.00	0.00	0.32	0.00	0.00
Methylcyclopentane	0.21	0.34	0.24	0.19	0.10	0.09	0.00	0.13	0.16	0.00	0.09
2,4-Dimethylpentane	0.51	1.08	0.37	0.14	0.09	0.05	0.00	0.20	0.16	0.00	0.06
Benzene	0.43	0.95	1.45	1.04	0.93	0.33	0.39	0.38	0.73	0.35	0.34
Carbon Tetrachloride	0.11	0.10	0.09	0.07	0.08	0.06	0.08	0.07	0.06	0.07	0.06
2-Methylhexane	7.18	15.8	6.43	3.50	2.49	1.39	0.47	2.32	1.77	0.10	0.22
3-methylhexane	7.39	16.0	5.88	1.86	1.48	0.73	0.56	2.41	1.95	0.12	0.24
Trichloroethene	0.27	0.23	0.20	0.31	0.21	0.13	0.08	0.09	0.46	0.13	0.22
Heptane	0.84	1.77	0.72	0.28	0.23	0.12	0.00	0.26	0.30	0.00	0.13
Methylcyclohexane	0.16	0.35	0.18	0.10	0.00	0.00	0.00	0.00	0.18	0.00	0.00
Toluene	1.76	2.07	1.35	6.31	13.7	1.81	21.0	3.26	29.3	8.89	101
Ethylbenzene	0.00	0.00	0.00	0.18	0.00	0.00	0.00	0.00	0.00	0.00	0.17
m-Xylene	0.00	0.00	0.11	0.31	0.00	0.00	0.00	0.00	0.14	0.00	0.34
o-Xylene	0.33	0.31	0.17	0.18	0.07	0.00	0.00	0.00	0.07	0.00	0.17
Chlorotoluenes	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
n-Propylbenzene	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.14	0.34
n-Decane	93.7	120	137	58.8	11.7	2.48	0.34	0.31	0.17	0.08	0.12
1,2,3-Trimethylbenzene	0.21	0.21	0.26	0.17	0.07	0.04	0.00	0.00	0.04	0.00	0.07
Undecane	0.31	0.29	0.62	0.29	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Naphthalene	0.20	0.00	0.11	0.41	0.22	0.00	0.00	0.00	0.00	0.00	0.00
Dodecane	1.08	1.58	2.02	1.35	1.93	0.35	0.11	0.16	0.14	0.15	0.14

Table 4. Targeted TFE and Telomer Alcohols – with Associated Area Counts

Target	TFE	4:2 fluorotelomer alcohol
Compounds	Area Counts	Area Counts
System Blank	1,363,994	0
Field Blank	209,616	1,601
Trip Blank	905,074	0
Ambient	767,244	0
Sample 1	3,263,269	8,816
Sample 2	1,635,088	24,170
Sample 3	1,112,791	14,648
Sample 4	1,510,977	62,473
Sample 5	791,857	70,496
Sample 6	1,360,932	38,139
Sample 7	1,504,011	106,613
Sample 8	2,361,816	20,520
Sample 9	1,220,128	0
Sample 10	2,610,103	0
Sample 11	2,621,143	0
Sample 12	15,207,595	0
Sample 13	13,819,373	9,319
Sample 14	13,280,328	10,148
Sample 15	8,559,907	0
Sample 16	13,565,851	0

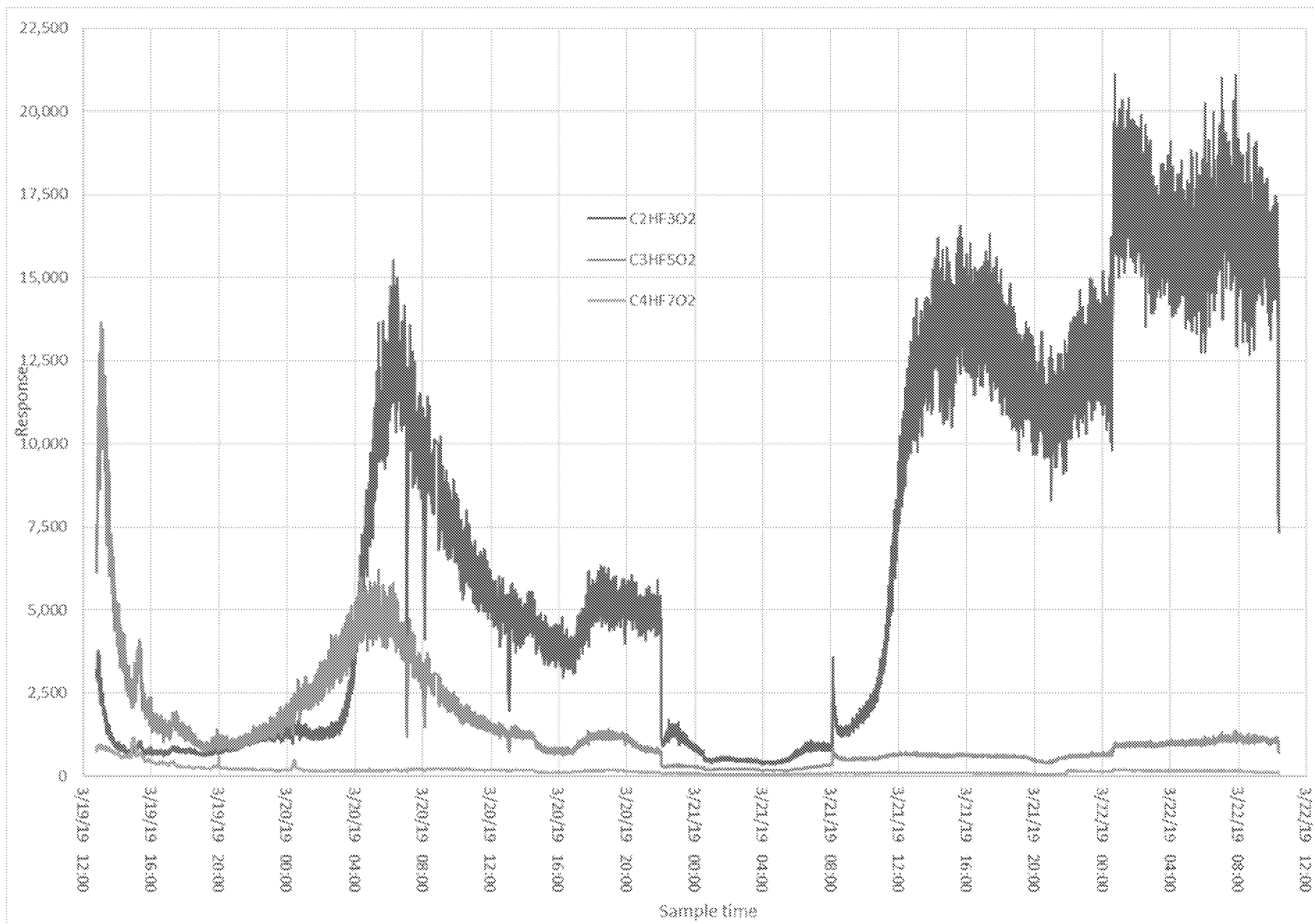


Figure 2. Graphically Depicted CIMS Results

Table 5. Non-Targeted PFAS Impinger Results – Compounds Identified with Associated Area Counts

	Reagent MeOH	Reagent Water	Proof Blank	Proof Blank	Field Blank	Field Blank 1	Field Blank 2	Nitrogen
			Train 1	Train 2	Train 1	Train 2	Train 2	System Blank
Formula	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts
C11H7F15O4	434	303	22,737	20,121	55,535	396,422	230,067	6,261
C13H9F17O4	471	495	823	811	1,116	219,924	204,015	2,420
C18H12F22O4	921	584	600	530	568	328,579	263,548	1,216
C15H11F19O4	252	159	247	226	339	111,135	145,063	579
C16H10F20O4	3,448	1,540	947	2,003	714	290,494	600,512	1,392
C9H5F13O4	572	4,244	615	4,240	1,018	364,699	93,143	2,545
C14H8F18O4	696	3,430	491	397	479	150,515	269,849	1,344
C12H13ClF12O7	479	452	997	805	2,235	62,007	69,606	2,349
C12H7F17O4	2,142	614	868	995	1,097	57,837	48,951	866
C12H4F20O7	313	258	2,788	459	1,907	318,115	744,483	5,453
C8H9F9N4O9S2	4,500	2,671	873	323	402	128,249	268,747	1,383
C11H5F13O4	324	402	898	1,074	868	26,207	36,585	1,341
C10H2F16O4	261	798	986	637	654	73,320	124,700	1,284
C14H7F17O4	472	382	501	412	611	72,474	122,861	1,487
C16H9F26O4P	12,442	874	277	470	616	191	1,378	215

Table 6. Train 2 Sample Results – Compounds Identified with Associated Area Counts

Tentative Identification	Neutral Mass	Formula	SAMPLE 1	SAMPLE 3	SAMPLE 5	SAMPLE 7	Field Blank 2	SAMPLE 9	SAMPLE 11	SAMPLE 13	SAMPLE 15
			Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts
Polyfluorinated Methyl Ester	488.01113	C11H7F15O4	6,657,458	2,897,711	3,040,281	935,850	230,067	322,222	326,127	775,940	425,894
Polyfluorinated Methyl Ester	552.02388	C13H9F17O4	6,102,196	1,699,905	1,661,338	614,575	204,015	152,918	238,232	447,115	183,283
Polyfluorinated Di-Acid	710.03979	C18H12F22O4	3,663,079	722,041	1,815,089	1,054,571	263,548	139,966	408,273	842,347	11,036
Polyfluorinated Methyl Ester	616.03641	C15H11F19O4	3,176,239	646,156	876,144	311,218	145,063	65,342	126,284	215,991	55,812
Polyfluorinated Di-Acid	646.02705	C16H10F20O4	2,957,655	1,636,440	1,424,529	877,979	600,512	134,151	408,419	1,242,858	63,947
Polyfluorinated Methyl Ester	423.99841	C9H5F13O4	1,606,702	2,010,592	1,282,317	569,953	93,143	264,123	191,428	432,503	379,980
Polyfluorinated Di-Acid	582.01444	C14H8F18O4	1,977,589	990,485	567,782	405,991	269,849	94,663	209,782	497,360	48,629
Polyfluorinated Methyl Ester	532.01765	C12H13ClF12O7	1,873,588	277,380	497,234	207,658	69,606	41,856	83,972	127,151	58,011
Polyfluorinated Di-Acid	538.00816	C12H7F17O4	1,554,081	412,882	336,178	153,453	48,951	39,058	59,029	115,767	55,580
Polyfluorinated Methyl Ester	639.96115	C12H4F20O7	1,449,245	413,398	914,355	590,579	744,483	132,756	373,496	680,681	8,784
Polyfluorinated Methyl Ester	539.96721	C8H9F9N4O9S2	1,348,053	316,192	486,826	288,797	268,747	68,459	196,222	481,852	18,974
Polyfluorinated Methyl Ester	447.99855	C11H5F13O4	1,231,393	75,124	116,148	71,855	36,585	15,300	49,264	120,815	21,962
Polyfluorinated Di-Acid	489.97023	C10H2F16O4	1,178,157	263,256	307,750	171,416	124,700	35,701	103,625	292,020	16,273
(3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-Heptafluorodecyl) hydrogen maleate	562.00837	C14H7F17O4	1,091,923	613,026	326,441	214,627	122,861	49,923	118,957	266,536	25,239
6:2 Fluorotelomer phosphate diester	789.98347	C16H9F26O4P	215	6,295	1,037	2,714	1,378	3,279	1,396	1,111	18,485

Table 7. Train 1 Sample Results – Compounds Identified with Associated Area Counts

			SAMPLE 2	SAMPLE 4	SAMPLE 6	SAMPLE 8	Field Blank	SAMPLE 10	SAMPLE 12	SAMPLE 14	SAMPLE 16
Tentative ID Name	Neutral Mass	Formula	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts	Area Counts
Polyfluorinated Methyl Ester	488.01113	C11H7F15O4	309	742	327	639	55,535	511	520	549	536
Polyfluorinated Methyl Ester	552.02388	C13H9F17O4	348	328	391	500	1,116	535	1,663	571	1,060
Polyfluorinated Di-Acid	710.03979	C18H12F22O4	658	440	558	511	568	1,034	423	441	527
Polyfluorinated Methyl Ester	616.03641	C15H11F19O4	207	209	176	206	339	332	275	241	426
Polyfluorinated Di-Acid	646.02705	C16H10F20O4	2,811	328	1,247	1,000	714	1,047	584	1,661	1,201
Polyfluorinated Methyl Ester	423.99841	C9H5F13O4	1,152	3,007	6,136	4,523	1,018	832	1,097	3,564	840
Polyfluorinated Di-Acid	582.01444	C14H8F18O4	738	207	367	3,753	479	672	1,672	599	602
Polyfluorinated Methyl Ester	532.01765	C12H13ClF12O7	579	262	387	461	2,235	500	1,603	469	871
Polyfluorinated Di-Acid	538.00816	C12H7F17O4	928	463	585	413	1,097	538	356	289	570
Polyfluorinated Methyl Ester	639.96115	C12H4F20O7	624	149	221	263	1,907	295	916	160	334
Polyfluorinated Methyl Ester	539.96721	C8H9F9N4O9S2	346	219	293	7,808	402	436	393	7,283	573
Polyfluorinated Methyl Ester	447.99855	C11H5F13O4	923	327	500	561	868	710	606	574	682
Polyfluorinated Di-Acid	489.97023	C10H2F16O4	795	296	409	489	654	673	522	617	608
(3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,10-Heptafluorodecyl) hydrogen maleate	562.00837	C14H7F17O4	437	298	379	738	611	685	613	486	614
6:2 Fluorotelomer phosphate diester	789.98347	C16H9F26O4P	562	947	1,598	2,466	616	3,520	632	2,369	243